First Asymmetry Synthesis and Stereochemistry of Glutinone

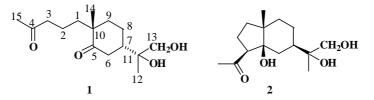
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Abstract: The first asymmetry synthesis of glutinone and its three diastereoisomers were carried out in 8 steps and the stereochemistry of this natural product was given out. **Keywords**: Asymmetry synthesis, asymmetry dihydroxylation, sesquiterpenoid.

Rencently, two new sesquiterpenoids, glutinone **1** and lucinone **2**, were isolated from the aerial parts of traditional medicine² *Jasonia glutinosa*¹ by Castillo and co-workers³. However, Castillo only gave the structure of glutinone, but its absolute configuration is still unknown. Here, we wish to report our stereoselective construction of the four possible diastereoisomers (**9a**, **9b**, **9c** & **9d**) of glutinone. By comparing the spectrum data with that of the natural product, we outlined the stereochemistry of glutione **1**.

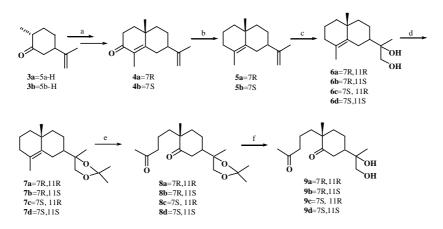
Scheme 1



Starting from (+)-dihydrocarvone **3a**, we got α -cyperone **4a** in three steps⁴. Deoxygenation⁵ of **4a** with AlCl₂H gave the diene **5a**. According to Sharpless' method⁶, **5a** was treated with commercially available AD-mix- β in *t*-BuOH-H₂O to afford diols **6a** in 90% yield. After **6a** was protected by its acetonide, which was oxidized by O₃ to give dione **8a**. Deprotection in 60% HOAc, one of the four possible diastereoisomers of glu- tione **9a** was obtained. Again, **9b** was obtained in the similar way when AD-mix- α was used. Next, same reactions were carried out from (-)-dihydrocarvone **3b**, we got the other two isomers **9c & 9d**. Confirmed by the spectrum data with that of the natural product, so we found that **9a** has the absolute configuration as shown for the natural compound $\mathbf{1}^7$.

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Scheme 2



Acknowledgments

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Reference and Notes

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- 5. For a review of the deoxygenation of carbonyl compound, see: S. Yamamura, S. Nishiyama, in *Comprehensive Organic Synthesis*, ed, B. M. Trost, I.Fleming, New York: Pergamon Press, **1991**, *8*, 307.
- 6. For a review of the asymmetry dihydroxylation, see: D. C. Kolb, M. S. Vannieuwenhze, K. B. Sharpless, *Chem. Rev.*, **1994**, *94*, 2483.
- 7. Compound **9a**: $[\alpha]_D^{17} = +64.3$ (c=1.0, CHCl₃); ¹H NMR (CDCl₃, 400M): δ ppm 1.11 (s, 3H, CH₃-14), 1.12 (s, 3H, CH₃-12), 2.11 (s, 3H, CH₃-15), 3.40, 3.55 (AB quartet, 2H, J=10.7 Hz, CH₂-13); ¹³C NMR (CDCl₃, 400M): δ ppm 18.28, 20.49, 22.34, 23.23, 30.01, 36.42, 37.34, 38.91, 44.35, 45.23, 47.39, 68.24, 73.90, 209.24, 215.77; MS (EI, 70ev), *m*/*z*: 252 (M⁺ H₂O).

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